K18U 1749

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Reg.	No.:	*****	 **********
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What is the radius of the orbital according to Bohr model ? V Semester B.Sc. Degree (CCSS – Supplementary) Examination, November 2018 (2011 & Earlier Admissions) are allowed as a factor Core Course in PHYSICS **5B10 PHY : Atomic, Nuclear & Particle Physics** the most stable and why

Total Weightage : 30 What is the spin angular momentum of a proton ?

Time: 3 Hours

9. The helium isotope offer is unsta

A - NOITCIES SECTION - A (Choose the correct answer. Each bunch carries a weightage of 1)

- 1. i) Balmer series contains wavelength in region.
 - d) microwave c) magnetic b) infrared a) visible
 - ii) When there are more atoms in the excited state than the ground state it is 11. Explain why the planetary model of the atom failed.
- s sent a) Laser sense tradonal edit in mean b) Optical pumping d) Metastable state c) Population inversion 13. What is symmetric and an symmetric
 - iii) Group 7 elements in periodic table are c) halogens d) non-metals b) metals a) inert gases
 - iv) In positive beta decay, _____ is emitted. In isomorphic meaning and the second sec c) pions and d) neutrons b) electron a) positron
- 2. i) When two nucleus join together to form a single nucleus, it emits energy. Distinguish between t
 - Such process is called b) nuclear fission a) weak interaction
 - Selemexe wd) binding energy line ere terlw. er c) nuclear fusion
 - ii) Hadrons consisting of three quarks are called c) baryons d) mesons b) tau neutrino a) leptons
 - iii) The coulomb energy of a nucleus is given by the most noted and a set c) -aA-1/3 d) -A a) -aA^{2/3} b) -aA
 - iv) Orbital angular momentum can acquire values b) $l\hbar$ c) $m\hbar$ d) ΣL_i a) h, to cosingl Obtain a relation for binding energy per nucleon through liquid drop model.

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SECTION - B

When there are more

(Answer any six questions. Each carries a weightage 1)

- 3. What is the radius of the orbital according to Bohr model ?
- 4. What is metastable state in lasers ?
- 5. What is an Pauli's exclusion principle ?
- (2011 & Earl) 6. Which element has a K_{α} X-ray line whose wavelength is 0.180nm ?
- 7. Which element is the most stable and why?
- 8. What is the spin angular momentum of a proton ?
- 9. The helium isotope ⁶₂He is unstable. What kind of decay would it undergo ?
- 10. Draw the diagram of common type of nuclear power plant ?

SECTION - C SW anishoo series remies

(Answer any nine questions. Each carries a weightage of two)

11. Explain why the planetary model of the atom failed.

- 12. What is the shortest wavelength present in the Brackett series of spectral lines ?
- 13. What is symmetric and antisymmetric wavefunction ? Group 7 elements in periodic table are
- 14. What is spin-orbit coupling ?
- 15. Explain semi-empirical mass formula.

16. Brief account of radioactivity.

- 17. Obtain a relation between the incident particles falling on a slab and its thickness.
- 18. Distinguish between the four fundamental interactions.
- 19. What are antimatters and give few examples ?

20. Explain about ITER.

- 21. Explain radioactive series.
- 22. Explain meson theory of nuclear forces. Usioun s to votene detailed and use and the

SECTION - D

(Answer any one question. Each carries a weightage of 4)

- 23. Obtain a relation for binding energy per nucleon through liquid drop model.
- 24. Explain nuclear fission and fusion.